

&oopmans

an open-source package for accurately predicting spectral properties

Edward Linscott | Theory and Simulation of Materials, EPFL | DFT2022, Brussels, 1 Sep 2022

Goal: spectral properties with a functional theory



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should be ...

- independent of its own occupation f_i
- equal to the corresponding total energy difference $E_i(N 1) E(N)$



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Differences to semi-local functionals:

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Differences to semi-local functionals:

• orbital-density dependence (can use MLWF densities)

$$\mathbf{v}_{i}^{\mathsf{KI}}/\alpha_{i} = -\mathbf{E}_{\mathsf{H}}\left[\mathbf{n}_{i}\right] + \mathbf{E}_{\mathsf{xc}}\left[\rho\right] - \mathbf{E}_{\mathsf{xc}}\left[\rho - \mathbf{n}_{i}\right] - \int d\mathbf{r}' \mathbf{v}_{\mathsf{xc}}(\mathbf{r}', [\rho]) \mathbf{n}_{i}(\mathbf{r}')$$

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Differences to semi-local functionals:

- orbital-density dependence (can use MLWF densities)
- screening

$$\frac{dE}{df_i} \approx \alpha_i \frac{\partial E}{\partial f_i}$$

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Differences to semi-local functionals:

- orbital-density dependence (can use MLWF densities)
- screening

$$\frac{dE}{df_i} \approx \alpha_i \frac{\partial E}{\partial f_i} \Rightarrow \varepsilon_i^{\text{Koopmans}} = \frac{\partial E_{\text{Koopmans}}}{\partial f_i} \approx E_i(N-1) - E(N)$$

Koopmans functionals: results for molecules

Ionisation potentials = $E(N - 1) - E(N) \stackrel{?}{=} -\varepsilon_{HO}$ of 100 molecules (the GW100 set) cf. CCSD(T)



Ultraviolet photoemission spectra



N. Colonna et al. J. Chem. Theory Comput. 14.5 (8, 2018), 2549; N. L. Nguyen et al. Phys. Rev. Lett. 114.16 (24, 2015), 166405



Mean absolute error (eV) across prototypical semiconductors and insulators

	PBE	G_0W_0	KI	KIPZ	QSGŴ
$E_{\rm gap}$	2.54	0.56	0.27	0.22	0.18
IP	1.09	0.39	0.19	0.21	0.49

N. L. Nguyen et al. Phys. Rev. X 8.2 (23, 2018), 021051

Koopmans functionals: results for solids



R. De Gennaro et al. Phys. Rev. B 106.3 (5, 2022), 035106

Koopmans functionals: results for solids



N. Colonna et al. J. Chem. Theory Comput. (4, 2022)

Screening coefficients $\{\alpha_i\}$ must be determined first, via...

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(a) finite difference calculations using a supercell



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(c) via machine learning (resulting band gaps within \sim 0.02 eV of explicit approach)

 $\tt kcw.x$ (DFPT implementation) is distributed in Quantum ESPRESSO v7.1 onwards

But complex workflows mean that...

- lots of different codes that need to handshake
- lots of scope for human error
- reproducibility becomes difficult
- expert knowledge required

Our solution...

&oopmans

- beta version just released¹
- implementations of Koopmans functionals
- automated workflows
 - start-to-finish Koopmans calculations
 - Wannierisation
 - dielectric tensor
 - convergence tests
 - ...
- built on top of ASE²
- does not require expert knowledge

koopmans-functionals.org



¹Linscott et al., in prep

² A. H. Larsen et al. J. Phys. Condens. Matter 29.27 (12, 2017), 273002

koopmans is scriptable

from ase.build import bulk from koopmans.kpoints import Kpoints from koopmans.projections import ProjectionBlocks from koopmans.workflows import SinglepointWorkflow

```
# Use ASE to create bulk silicon
atoms = bulk('Si')
```

```
# Create the workflow
workflow = SinglepointWorkflow(atoms = atoms,
    projections = si_projs,
    ecutwfc = 40.0,
    kpoints = Kpoints(grid=[8, 8, 8], path='LGXKG', cell=atoms.cell),
    calculator_parameters = {'pw': {'nbnd': 10},
    'w90_emp': {'dis_froz_max': 10.6, 'dis_win_max': 16.9}})
```

```
# Run the workflow
workflow.run()
```

Take home messages



- Koopmans functionals are a class of functionals that treat spectral properties on the same footing as total energy differences (via GPWL)
- they can give orbital energies and band structures with comparable accuracy to state-of-the-art GW
- the release of koopmans means you don't need expert knowledge to run Koopmans functional calculations

Want to find out more? Go to koopmans-functionals.org Free online school Nov 9-11 2022 Advanced Quantum ESPRESSO tutorial: Hubbard and Koopmans functionals from linear response. Register at https://sites.google.com/view/hubbard-koopmans/home Follow 🛫 @ed_linscott for updates | Slides available at elinscott.github.io



SPARE SLIDES

Recap from earlier

Key idea: construct a functional such that the variational orbital energies

$$\varepsilon_i^{\mathsf{Koopmans}} = \langle \varphi_i | \mathcal{H} | \varphi_i \rangle = \partial \mathcal{E}_{\mathsf{Koopmans}} / \partial f_i$$

are...

- independent of the corresponding occupancies *f_i*
- equal to the corresponding total energy difference $E_i(N-1) E(N)$

zero band gap \rightarrow occupancy matrix for variational orbitals is off-diagonal

Learning the screening parameters



Y. Schubert et al. "Predicting Screening Parameters For Fast Koopmans Spectral Functional Calculations". 2022

$$c_{nlm,k=\text{orbital}}^{i} = \int d\mathbf{r} g_{nl}(\mathbf{r}) Y_{lm}(\theta,\varphi) \rho^{i}(\mathbf{r}-\mathbf{R}^{i})$$

 g_{nl} = orthonormalised radial Gaussian basis functions Y_{lm} = spherical harmonics

$$p_{n_1n_2l,k_1k_2}^i = \pi \sqrt{\frac{8}{2l+1}} \sum_m c_{n_1lm,k_1}^{i*} c_{n_2lm,k_2}^i$$

Learning the screening parameters



loss of accuracy of the band gap of \sim 0.02 eV (cf. when calculating screening parameters *ab initio*)

Consequences of ODD

- a natural generalisation in the direction of spectral functional theory¹
- variational (localised, minimising) vs canonical (delocalised, diagonalising) orbitals



(a) variational

(b) canonical

- ODD functional means that we know Ĥ|φ_i⟩ for variational orbitals {|φ_i⟩} but we don't know Ĥ in general
- Difficulties when it comes to calculating transport properties/spectra
- Perhaps a DFT+U-projector approach is more convenient?

¹ A. Ferretti et al. *Phys. Rev. B* 89.19 (27, 2014), 195134.

N. L. Nguyen et al. Phys. Rev. X 8.2 (23, 2018), 021051

Dabo, I. et al. *Phys. Rev. B* 82.11 (23, 2010). Borghi, G. et al. *Phys. Rev. B* 90.7 (20, 2014). Colonna, N. et al. J. Chem. Theory Comput. 15.3 (12, 2019). - .J. Chem. Theory Comput. 14.5 (8, 2018). Nguyen, N. L. et al. Phys. Rev. Lett. 114.16 (24, 2015). Nguyen, N. L. et al. *Phys. Rev. X* 8.2 (23, 2018). De Gennaro, R. et al. Phys. Rev. B 106.3 (5, 2022). Colonna, N. et al. J. Chem. Theory Comput. (4, 2022). Schubert, Y. et al. 2022. Larsen, A. H. et al. J. Phys. Condens. Matter 29.27 (12, 2017). Ferretti, A. et al. Phys. Rev. B 89.19 (27, 2014).